Chem 322	(Sturgeon)
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Name

Photoelectric effect

1) What is the speed/velocity of the ejected photoelectron when 330 nm light is shown on the surface of lithium?

(where might someone find the work function for lithium???)

Answer: 5.51e5 m/s

2a) The following set of data were observed in an experiment on the photoelectric effect from potassium (ϕ = 2.26 eV). Make an Excel spreadsheet for this data, then generate a graph (Freq vs KE) similar to the figure in the text (4th Edition – Fig 1.4).

Frequency	Velocity	Kinetic Energy (J)	Kinetic Energy (eV)
(1/s)	(m/s)		
	9.93E+05		
	8.24E+05		
	6.44E+05		
	5.42E+05		
	3.92E+05		
	2.61E+05		
		(1/s) (m/s) 9.93E+05 8.24E+05 6.44E+05 5.42E+05 3.92E+05	(1/s) (m/s) 9.93E+05 8.24E+05 6.44E+05 5.42E+05 3.92E+05

Work function = "global variable" List of constants = "global variables"

<<attached spreadsheet>>

2b) Using the data above, what is the value for "h" (Plank's Constant)?