

Potpourri

~~Discussion~~

10:00 - 11:30 a.m.

Oct. 24

The <sup>entitled</sup> discussion group <sup>✓</sup> Potpourri numbered 10 enterprised souls who grappled with issues ranging from "How do we know what we know in science?" to "How do we explain the fact that air filled soap bubbles expand ~~in the process~~ as when they move from a warm air environment to a cold CO<sub>2</sub> environment found in the neighborhood of dry ice?"

~~Non-scientists courses~~

The rewards, trials and tribulations, and test book problems related to teaching courses for "non-scientists", particularly physical science courses and <sup>Chemistry</sup> ✓ courses for elementary education majors, were ~~shared~~ discussed.

Experiences ~~shared~~ with courses that

explore the impact of science on the human condition were <sup>shared</sup> ~~discussed~~ by two or three individuals in the group.

The issue of the faculty member as Professor (professional in his subject matter) vs the ~~the~~ faculty member as teacher (in the sense of being non-professional in the subject matter) was briefly debated.

Finally Dick Ramette left us all scratching our heads trying to explain on the basis of molecular behavior how it is that  $\sqrt{\frac{1}{2}}$  ~~at~~ helium balloons <sup>a</sup> and carbon dioxide filled balloons, tethered inside an automobile, respond differently when the automobile is accelerated.

- Charles Carlin Recorder

Irene Break Caver