Ch 4. Precipitation Reactions Practice (Chem 140)  Name
Write out the following precipitation chemical reactions, the ionic equation, and the net ionic equation. Balance and include physical states of all reactants and products.
1) Potassium phosphate reacting with calcium nitrate
Rxn:
Ionic eq:
Net ionic eq:
2) Silver nitrate reacting with sodium chloride
Rxn:
Ionic eq:
Net ionic eq:
3) Strontium fluoride reacting with lithium sulfate
Rxn:
Ionic eq:
Net ionic eq:
4) Sodium carbonate reacting with magnesium iodide
Rxn:
Ionic eq:
Net ionic eq:
5) Potassium hydroxide reacting with calcium bromide
Rxn:
Ionic eq:

Net ionic eq: